

OCURRENCE OF 1,2,4-TRIAZOLE  
RING IN ACTINOMYCETES

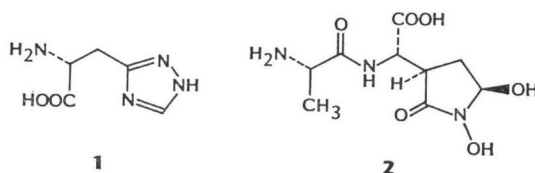
Sir:

In the course of our screening for new anti-metabolites from actinomycetes, it was found that strain KM-10329 produced L-1,2,4-triazole-3-alanine. Many synthetic compounds containing the 1H-1,2,4-triazole ring are known, however, their natural occurrence has never previously been reported. In this present communication, the isolation and identification of the compound are described.

Strain KM-10329 was isolated from a soil sample collected at Joetsu-city, Niigata Prefecture, Japan. On the basis of taxonomic studies, it was classified as *Streptomyces* sp.

This organism was cultured at 27°C for 48 hours in 500-ml Erlenmeyer flasks containing 100 ml of a seed medium, composed of glucose 0.1%, starch 2.4%, peptone 0.3%, meat extract 0.3%, yeast extract 0.5% and CaCO<sub>3</sub> 0.4% (pH 7.0). Two hundred milliliters of the seed culture were transferred into 20 liters of a medium containing wheat bran 4.0%, soybean meal 2.0% and NaCl 0.3% (pH 7.0) in a 30-liter jar fermentor, and the fermentation was carried out at 27°C for 140 hours under aerobic conditions.

The culture filtrate (15 liters) was applied to a column of Amberlite IR-120B (H<sup>+</sup>). The column was washed with H<sub>2</sub>O, and then the active material was eluted with 1.0 N NH<sub>4</sub>OH. The eluate was concentrated *in vacuo*, and applied to a Diaion PA-416 (OH<sup>-</sup>) column. After washing the column with H<sub>2</sub>O, the active principle was eluted with 0.1 N AcOH. The active eluate was concentrated to a small volume *in vacuo*, and poured into MeOH. After the flocculated material was removed by filtration, the filtrate was evaporated to dryness, and chromatographed on an Avicel column with EtOH-0.15 M aq NH<sub>4</sub>OH. The active fractions were collected, concentrated *in vacuo*, and lyophilized to yield a brown powder. The powder was dissolved in a small amount of H<sub>2</sub>O and purified on an activated carbon column with 10% MeOH in H<sub>2</sub>O. Acetone was added to the concentrated active eluate to yield crystalline 1: 150 mg; mp 258~261°C (dec);  $\lambda_{\text{max}}^{\text{MeOH}}$  nm 255 (sh); IR (KBr) 2850, 1605, 1525, 1430, 1410, 1020 cm<sup>-1</sup>; EIMS *m/z* 157 (M+H)<sup>+</sup>, HIMS Calcd for C<sub>5</sub>H<sub>8</sub>N<sub>4</sub>O<sub>2</sub>: 157.073; Found: 157.073;



*Anal* Calcd for C<sub>5</sub>H<sub>8</sub>N<sub>4</sub>O<sub>2</sub>: C 38.46, H 5.16, N 35.88; Found C 38.15, H 5.05, N 34.86; [ $\alpha$ ]<sub>D</sub><sup>15</sup> -18.0° (c 1.0, H<sub>2</sub>O).

The <sup>1</sup>H NMR spectrum of **1** in D<sub>2</sub>O showed one methylene [3.28 (d, *J*=7.6 Hz) and 3.31 (d, *J*=5.4 Hz) ppm] coupled with a methine [4.08 (1H, dd, *J*=7.6 and 5.4 Hz) ppm] which corresponded to the α-proton of α-amino acid, and an isolated olefinic proton [8.28 (1H, s) ppm]. These data suggested the presence of a β-substituted alanine moiety. The <sup>13</sup>C NMR spectrum of **1** in D<sub>2</sub>O showed 5 carbon signals, *i.e.*, one methylene (28.9 ppm, <sup>1</sup>*J*<sub>CH</sub>=130.4 Hz), a methine (54.1 ppm, <sup>1</sup>*J*<sub>CH</sub>=145.3 Hz), a protonated and a non-protonated *sp*<sup>2</sup> carbon (146.9 ppm, <sup>1</sup>*J*<sub>CH</sub>=212.7 Hz and 157.3 ppm), and a carboxylic acid (173.6 ppm). The large one bond <sup>13</sup>C-<sup>1</sup>H coupling constant of the protonated *sp*<sup>2</sup> carbon indicated that the carbon was between two nitrogens.<sup>1)</sup> Since the remaining carbon had to attach to nitrogen through a double bond and the <sup>13</sup>C NMR chemical shift of methylene was similar with that of the β-carbon of histidine, the structure of **1** was concluded to be 1,2,4-triazole-3-alanine. In order to confirm this, the <sup>1</sup>H NMR spectrum of **1** was compared with that of authentic D,L-1,2,4-triazole-3-alanine, and these spectra were superimposable.

The specific optical rotation of **1** for D line was -18.0° and that of L-histidine was known to be -39.7°, consequently the stereochemistry of **1** is the L form. Furthermore, positive Cotton effect in CD curve was observed both in **1** and L-histidine, confirming that they are L form. Therefore, the structure of **1** is L-1,2,4-triazole-3-alanine. The synthesized D,L-1,2,4-triazole-3-alanine has been known as a histidine antagonist,<sup>2)</sup> and the mechanism of action has been investigated.<sup>3)</sup>

The strain was also found to co-produce a proline antagonist,<sup>4)</sup> nourseimycin (**2**), having collagen proline hydroxylase inhibiting activity.<sup>5)</sup> The compound also contains an unusual amino acid.<sup>6)</sup> Thus, the strain KM-10329 produces two unusual amino acids.

This is the first report of the natural occurrence of a compound containing 1,2,4-triazole ring. There are no structural differences except that in **1** nitrogen is substituted for the  $\delta$ -carbon in L-histidine. Biosynthesis of L-histidine has been well investigated, *i.e.*, the first step is the condensation of ATP with phosphoribosylpyrophosphate, and the  $\delta$ -carbon of L-histidine is derived from the anomeric carbon of ribose. Though the structure of **1** is closely related to L-histidine, the biosynthesis of **1** seems not to relate with that of L-histidine. Thus, it is an interesting target for biosynthetic investigation.

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